

1. Generic Matrix

Element of examination	Competence	Weight in %	Evaluation/Marking	Weight in marks
Fields	Knowledge and Comprehension	$\pm 37\%$	Paper-specific Marking Scheme	± 11
	Application	$\pm 37\%$		± 11
	Analysis and Evaluation	$\pm 16\%$		± 5
	Written Communication	$\pm 10\%$		± 3
		100 %		30
Waves	Knowledge and Comprehension	$\pm 37\%$	Paper-specific Marking Scheme	± 11
	Application	$\pm 37\%$		± 11
	Analysis and Evaluation	$\pm 16\%$		± 5
	Written Communication	$\pm 10\%$		± 3
		100 %		30
Atomic Physics	Knowledge and Comprehension	$\pm 35\%$	Paper-specific Marking Scheme	± 7
	Application	$\pm 35\%$		± 7
	Analysis and Evaluation	$\pm 20\%$		± 4
	Written Communication	$\pm 10\%$		± 2
		100 %		20
Nuclear Physics	Knowledge and Comprehension	$\pm 35\%$	Paper-specific Marking Scheme	± 7
	Application	$\pm 35\%$		± 7
	Analysis and Evaluation	$\pm 20\%$		± 4
	Written Communication	$\pm 10\%$		± 2
		100 %		20
Total Exam				100

In each section a deviation of up to 5% will be tolerated as long as the total number of marks (30 points respectively 20 points) is respected for each question.

2. Paper-specific Matrix

Element of examination	Competence	Weight in %	Evaluation/Marking	Weight in marks
Fields	Knowledge and Comprehension	35.0 %	Paper-specific Marking Scheme see part 4	10.5
	Application	38.3 %		11.5
	Analysis and Evaluation	18.3 %		5.5
	Written Communication	8.3 %		2.5
		100 %		30
Waves	Knowledge and Comprehension	40.0 %	Paper-specific Marking Scheme see part 4	12.0
	Application	33.3 %		10.0
	Analysis and Evaluation	16.7 %		5.0
	Written Communication	10.0 %		3.0
		100 %		30
Atomic Physics	Knowledge and Comprehension	32.5 %	Paper-specific Marking Scheme see part 4	6.5
	Application	37.5 %		7.5
	Analysis and Evaluation	20.0 %		4.0
	Written Communication	10.0 %		2.0
		100 %		20
Nuclear Physics	Knowledge and Comprehension	35.0 %	Paper-specific Marking Scheme see part 4	7.0
	Application	35.0 %		7.0
	Analysis and Evaluation	17.5 %		3.5
	Written Communication	12.5 %		2.5
		100 %		20
Total Exam				
				100

3. Sample BAC Written Examination

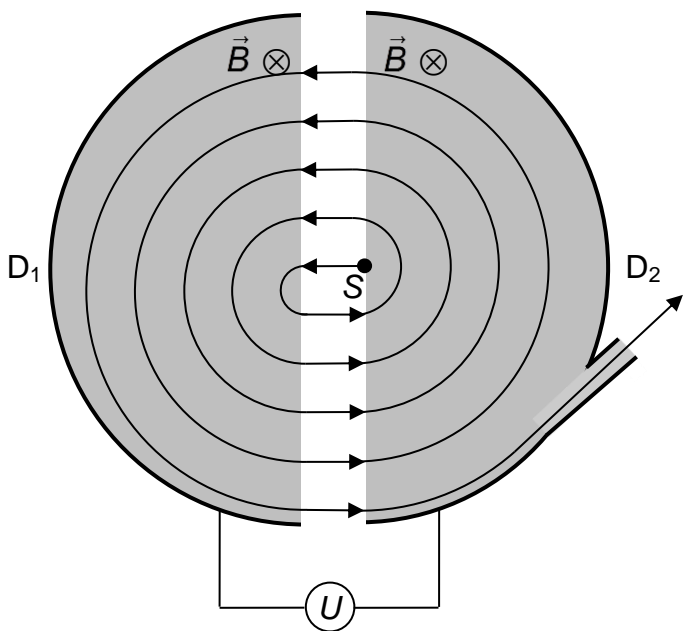
Question 1																																						
Part A	Page 1/4	Marks																																				
<p>TRAPPIST-1 is an ultra-cool red dwarf star that is slightly larger, but has a much greater mass than the planet Jupiter. On February 22nd, 2018, astronomers announced that the planetary system of this star is composed of seven planets.</p> <p>In this question, assume that all planets are moving in circular orbits.</p>																																						
<table border="1" style="margin: auto; border-collapse: collapse;"> <thead> <tr> <th colspan="4">The TRAPPIST-1 planetary system</th> </tr> <tr> <th style="width: 15%;">Planet</th> <th style="width: 25%;">Mass (Earth masses)</th> <th style="width: 25%;">Orbital radius (10⁶ km)</th> <th style="width: 35%;">Orbital period (Earth days)</th> </tr> </thead> <tbody> <tr><td>b</td><td>1.02</td><td>1.73</td><td>1.51</td></tr> <tr><td>c</td><td>1.16</td><td>2.37</td><td>2.42</td></tr> <tr><td>d</td><td>0.30</td><td>3.33</td><td>4.05</td></tr> <tr><td>e</td><td>0.77</td><td>4.38</td><td>6.10</td></tr> <tr><td>f</td><td>0.93</td><td>5.76</td><td>9.21</td></tr> <tr><td>g</td><td>1.14</td><td>7.01</td><td>12.35</td></tr> <tr><td>h</td><td>0.33</td><td>9.27</td><td>18.77</td></tr> </tbody> </table> <p style="font-size: small; margin-top: 5px;">Source: Wikipedia EN, Jan 18th, 2019</p>			The TRAPPIST-1 planetary system				Planet	Mass (Earth masses)	Orbital radius (10 ⁶ km)	Orbital period (Earth days)	b	1.02	1.73	1.51	c	1.16	2.37	2.42	d	0.30	3.33	4.05	e	0.77	4.38	6.10	f	0.93	5.76	9.21	g	1.14	7.01	12.35	h	0.33	9.27	18.77
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<p>a) Kepler's 3rd law states that for planetary orbits $\frac{T^2}{r^3}$ is a constant, where T is the orbital period and r the orbital radius.</p> <p>Verify Kepler's 3rd law using data for 2 planets from the table above.</p>		3 marks																																				
<p>b) Show that the orbital velocity of planet "e" is equal to $v_e = 5.22 \times 10^4 \text{ m s}^{-1}$.</p>		3 marks																																				
<p>c) For any two planets that are orbiting at a distance r_1 and r_2 from a star, the ratio of their orbital velocities v_1 and v_2 is given by</p> $\frac{v_1}{v_2} = \sqrt{\frac{r_2}{r_1}} .$ <p>Derive this expression.</p>		3 marks																																				
<p>d) One of the planets of TRAPPIST-1 has an orbital velocity of $4.13 \times 10^4 \text{ m s}^{-1}$.</p> <p>Which planet is it?</p>		3 marks																																				

3. Sample BAC Written Examination

Question 1		
Part A	Page 2/4	Marks
<p>e) i. Show that the total mechanical energy of a planet orbiting around a star is given by</p> $E_{\text{tot}} = -G \frac{mM}{2r},$ <p>where m is the mass of the planet, M the mass of the star, and r the distance between the planet and the star.</p> <p>ii. The mass of TRAPPIST-1 is 1.77×10^{29} kg. Calculate the total mechanical energy of planet “e”.</p>		<p>3 marks</p> <p>1 mark</p>

Part A	
<p><u>Given:</u></p> <p>mass of the Earth</p> <p>universal gravitational constant</p>	$m_E = 5.97 \times 10^{24} \text{ kg}$ $G = 6.67 \times 10^{-11} \text{ m}^3 \text{ kg}^{-1} \text{ s}^{-2}$

3. Sample BAC Written Examination

Question 1		
Part B	Page 3/4	Marks
<p>A cyclotron is a particle accelerator. It consists of two hollow half-cylinders D_1 and D_2, known as Dees, separated by a narrow gap (see figure below).</p> <p>In an experiment, protons are emitted with negligible initial velocity by the source S.</p> <p>In the gap between the Dees, the protons are accelerated by a potential difference U. The potential difference changes sign after every passage of the protons through the gap. The absolute value of this potential difference is $U = 1.00 \times 10^4 \text{ V}$ when a proton crosses the gap.</p> <p>A uniform magnetic field \vec{B} with $B = 1.00 \text{ T}$ is present inside the Dees, with a direction parallel to the axis of the half-cylinders.</p> <p>The subsequent trajectory of the protons in each Dee is circular. The radius increases after each crossing through the gap.</p>		
		

3. Sample BAC Written Examination

Question 1		
Part B	Page 4/4	Marks
<p>a) A proton enters a Dee with speed v.</p> <p style="padding-left: 20px;">i. Show that the radius R of its trajectory is given by:</p> $R = \frac{m_p v}{e B}$ <p style="padding-left: 20px;">ii. Show, by deriving a formula for the time Δt spent in a Dee, that this time is independent of speed.</p>		<p>3 marks</p> <p>2 marks</p>
<p>b)</p> <p style="padding-left: 20px;">i. Show that the increase in kinetic energy of a proton for each gap crossing is 1.00×10^4 eV .</p> <p style="padding-left: 20px;">ii. Calculate the radius R_1 of the first circular trajectory.</p>		<p>2 marks</p> <p>3 marks</p>
<p>c) A proton accelerated by the cyclotron has its maximum energy as it exits the Dee after its last revolution. The radius of the trajectory at the exit from the cyclotron is $R_{max} = 0.289$ m .</p> <p style="padding-left: 20px;">i. Show that the maximum kinetic energy of this proton is $E_{max} = 4.00$ MeV .</p> <p style="padding-left: 20px;">ii. Calculate the number of revolutions necessary for this proton to acquire the maximum kinetic energy.</p>		<p>3 marks</p> <p>1 mark</p>

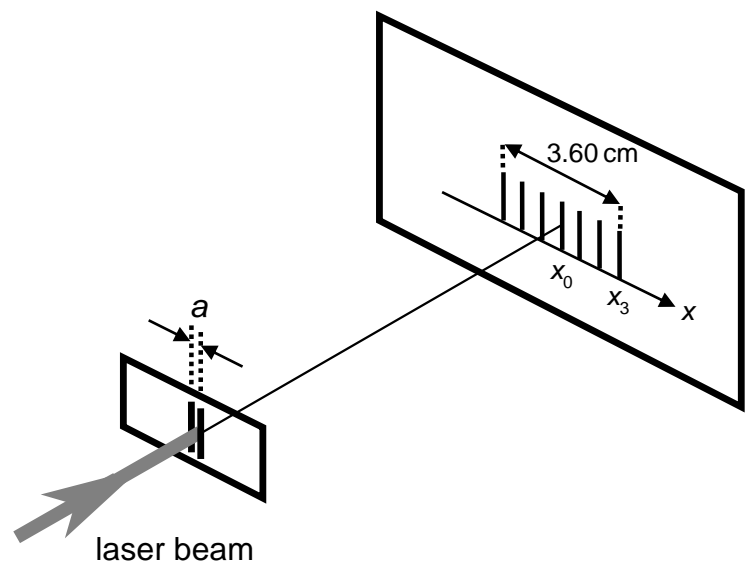
Part B	
<u>Given:</u>	
elementary charge	$e = 1.60 \times 10^{-19}$ C
mass of the proton	$m_p = 1.67 \times 10^{-27}$ kg

3. Sample BAC Written Examination

Question 2		
Part A	Page 1/3	Marks
<p>The lengths of organ pipes vary between several metres and a few centimetres. Some of the pipes are open at both ends (“open pipes”) and others are open at one end and closed at the other end (“closed pipes”).</p> <p>The human ear can hear sounds with frequencies between 20 Hz and 16 000 Hz.</p>		
a)	<p>i. For both types of pipes, sketch diagrams of the fundamental wave and the first overtone, indicating the position of the nodes.</p>	4 marks
	<p>ii. Calculate the lengths of both types of pipes which produce a fundamental note of 20 Hz.</p>	3 marks
	<p>iii. For two pipes with the same length, one “open” and one “closed”, calculate the ratio of the frequencies of their first overtones.</p>	2 marks
b)	<p>Consider a note of frequency 440 Hz. If you go down or up one octave the frequency either halves or doubles respectively.</p>	
	<p>i. Calculate the frequency of a note, which is four octaves below 440 Hz, and decide whether the human ear can still hear that note.</p>	2 marks
	<p>ii. The frequency of the highest note several octaves above 440 Hz which we can still hear is 14080 Hz.</p>	
	<p>1. Calculate how many octaves it lies above the 440 Hz.</p>	1 mark
	<p>2. The shortest pipe in an organ is 6.14 mm long.</p> <p>Decide by using a calculation whether it is an “open pipe” or a “closed pipe”, knowing that its fundamental frequency is 14080 Hz.</p>	3 marks

Part A	
<p>Given:</p> <p>speed of sound in air</p>	$v_{\text{sound}} = 346 \text{ m s}^{-1}$

3. Sample BAC Written Examination

Question 2		
Part B	Page 2/3	Marks
<p>a) Students perform Young's experiment using laser light with wavelength λ. The light is incident on a double slit with slit separation a. An interference pattern is observed on a screen located at the distance L from the double slit. The screen is parallel to the plane of the double slits.</p> <p>i. Show that the positions of the maxima on the screen is given by:</p> $x_k = k \frac{L \lambda}{a}, \text{ where } k = 0, \pm 1, \pm 2, \dots$ <p>State the approximations used.</p> <p>ii. Knowing that the distance between the two 3rd order maxima on the screen is 3.60 cm, $L = 4.00$ m and $\lambda = 546$ nm, calculate the slit separation a (see figure below).</p>  <p>The diagram illustrates the experimental setup. A laser beam is directed at a double slit with slit separation a. The distance to the screen is L. The screen shows an interference pattern with maxima at positions x_0 and x_3. The distance between x_0 and x_3 is labeled as 3.60 cm.</p>		<p>4 marks</p> <p>2 marks</p>
<p>b) Using a double slit with $a = 3.64 \times 10^{-4}$ m, the students replace the laser with a source, which emits red light ($\lambda_1 = 672$ nm) and green light (λ_2). On the screen there is an overlap of interference patterns. A maximum for green overlaps with the third order maximum for red.</p> <p>Determine the wavelength λ_2 of the green light and the order of the green maximum overlapping the red maximum.</p>		<p>4 marks</p>

3. Sample BAC Written Examination

Question 2		
Part B	Page 3/3	Marks
<p>c) Students use another laser and replace the double slit with a diffraction grating with 4000 lines per centimetre. The distance $L = 4.00$ m remains unchanged. The first maximum is observed at a distance 0.871 m from the central maximum on the screen. The diffraction grating formula is</p> $k\lambda = d\sin(\theta_k) .$ <p>i. Explain the meaning of d and θ_k in this formula.</p> <p>ii. Show that the wavelength of the laser light is 532 nm.</p>		<p>1 mark</p> <p>4 marks</p>

Part B	
<u>Given:</u>	
wavelength of green light	$500 \text{ nm} \leq \lambda \leq 560 \text{ nm}$
speed of light in a vacuum	$c = 3.00 \times 10^8 \text{ m s}^{-1}$

3. Sample BAC Written Examination

Question 3														
	Page 1/1	Marks												
<p>a) The equation below is Einstein's equation describing the photoelectric effect when a photocell is illuminated by light of frequency f</p> $hf = W_0 + E_{\text{kin}}$ <p>i. Explain what is meant by the three terms hf, W_0 and E_{kin}.</p> <p>ii. Monochromatic light of wavelength 486 nm is used to illuminate the photocell. The photocathode is coated with a thin layer of caesium with a work function of 2.08 eV and has a surface area of 100 mm².</p> <p>The intensity of the light which is incident on the cathode of the photoelectric cell is 0.100 W m⁻².</p> <ol style="list-style-type: none">Show that the energy of a single photon of this light is 4.09×10^{-19} J.Calculate the maximum kinetic energy of a photoelectron.Show that the number of photons incident on the surface of the photocathode equals 2.44×10^{13} per second.Calculate the maximum photocurrent assuming that 4 % of the photons result in photoelectron emission.		3 marks												
<p>b) In the hydrogen atom spectrum the wavelengths can be sorted into series like the Balmer series.</p> <p>The photons of the Balmer series are emitted when electrons make transitions from states with the quantum number $n \geq 3$ to $n = 2$. The table below shows the values of the first five energy levels E_n for the hydrogen atom.</p> <table border="1"><thead><tr><th>Quantum number n</th><th>1</th><th>2</th><th>3</th><th>4</th><th>5</th></tr></thead><tbody><tr><td>E_n / eV</td><td>-13.6</td><td>-3.40</td><td>-1.51</td><td>-0.85</td><td>-0.54</td></tr></tbody></table> <p>One of the Balmer series transitions results in the emission of a photon with wavelength 486 nm.</p> <p>Between which energy levels does this transition occur?</p>	Quantum number n	1	2	3	4	5	E_n / eV	-13.6	-3.40	-1.51	-0.85	-0.54		4 marks
Quantum number n	1	2	3	4	5									
E_n / eV	-13.6	-3.40	-1.51	-0.85	-0.54									

Given:

Planck's constant $h = 6.63 \times 10^{-34}$ J s

speed of light in a vacuum $c = 3.00 \times 10^8$ m s⁻¹

elementary charge $e = 1.60 \times 10^{-19}$ C

3. Sample BAC Written Examination

Question 4		
	Page 1/2	Marks
<p>a) One of the isotopes of the element technetium is ${}_{43}^{99}\text{Tc}$.</p> <p>i. What do we mean by the term “isotope”?</p> <p>ii. What is the composition of the nucleus of this isotope?</p> <p>iii. ${}_{43}^{99}\text{Tc}$ decays into ${}_{44}^{99}\text{Ru}$.</p> <p>Write the decay equation of ${}_{43}^{99}\text{Tc}$ to ${}_{44}^{99}\text{Ru}$ and state which type of decay it is.</p> <p>Technetium-99m is a metastable isotope that decays to ${}_{43}^{99}\text{Tc}$ by emitting gamma radiation. Technetium-99m is widely used in nuclear medicine. In the graph below you can see the activity of a sample of technetium-99m:</p> <div style="text-align: center;"> </div>	<p>1 mark</p> <p>1 mark</p> <p>2 marks</p>	
<p>iv. Explain the meaning of the term “half-life”, $T_{1/2}$, for a radioactive isotope.</p> <p>v. From the graph, estimate the half-life of technetium-99m.</p> <p>vi. Show that $T_{1/2} = \frac{\ln(2)}{\lambda}$, where λ is the decay constant.</p>	<p>1 mark</p> <p>1 mark</p> <p>2 marks</p>	

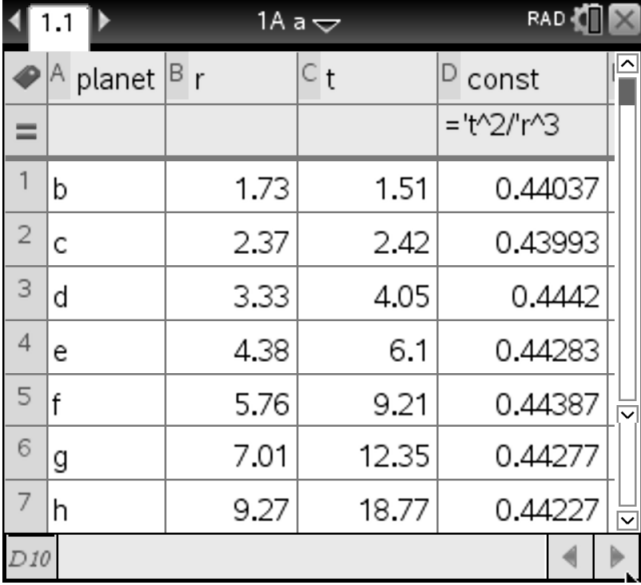
3. Sample BAC Written Examination

Question 4		
	Page 2/2	Marks
<p>b) One of the fission reactions that uranium may undergo in a nuclear reactor is:</p> ${}_{92}^{235}\text{U} + {}_0^1\text{n} \rightarrow {}_{56}^{141}\text{Ba} + {}_{36}^{92}\text{Kr} + 3{}_0^1\text{n}$		
<p>i. Explain how a chain reaction is produced, and the role of a moderator in a nuclear reactor.</p>		4 marks
<p>ii. Calculate the energy released by this reaction.</p>		4 marks
<p>c) In a reactor using uranium-235, a variety of fission reactions occur. The average energy released per fission is 210 MeV.</p> <p>Calculate the mass of uranium-235 that undergoes fission per hour to run a 2.00 GW power plant assuming an efficiency of 33 %.</p>		4 marks

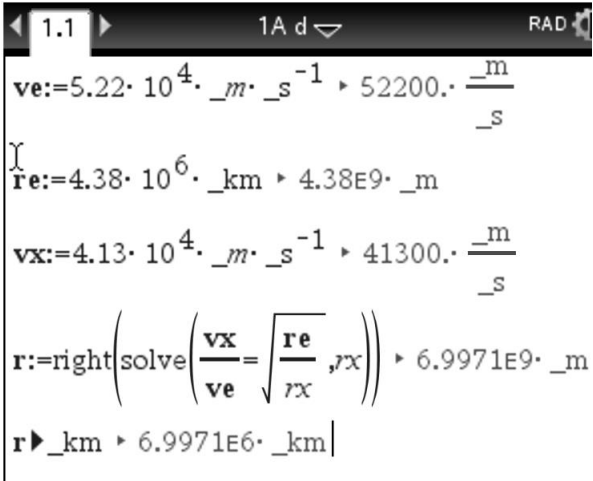
Given:

atomic mass unit	$1 \text{ u} = 931.5 \text{ MeV} / c^2 = 1.66 \times 10^{-27} \text{ kg}$
speed of light in a vacuum	$c = 3.00 \times 10^8 \text{ m s}^{-1}$
elementary charge	$e = 1.60 \times 10^{-19} \text{ C}$
mass of the neutron	$m_n = 1.008\,665 \text{ u}$
atomic mass of ${}_{36}^{92}\text{Kr}$	91.926 156 u
atomic mass of ${}_{56}^{141}\text{Ba}$	140.914 411 u
atomic mass of ${}_{92}^{235}\text{U}$	235.043 930 u

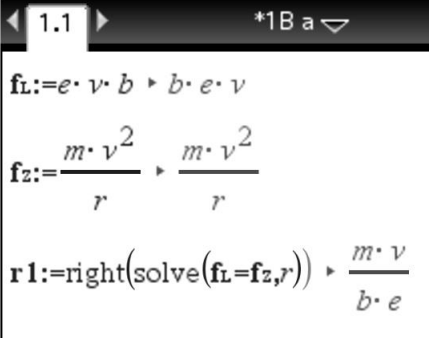
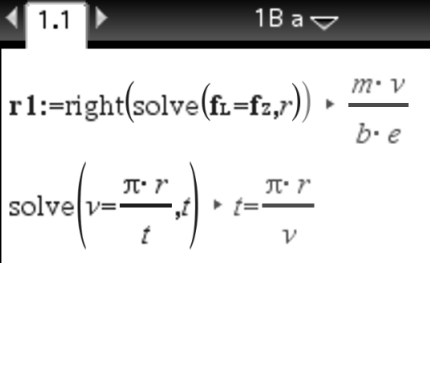
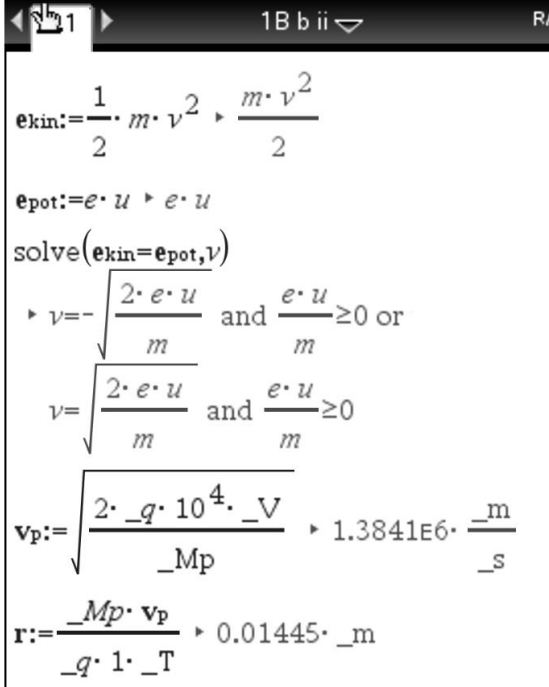
4. Marking Scheme and 5. Answer Sheet

Solution to Question 1, Part A		Fields				
A: Knowledge and Comprehension; B: Application; C: Analysis and Evaluation; W: Written Communication		A	B	C	W	Σ
a)	 <p>Kepler's 3rd law gives (nearly) the same value of the ratio $\frac{T^2}{r^3}$ for all the planets. (As stated in the question, this ratio needs to be calculated and shown to be constant for only two planets.)</p>	2				3
b)	$v_e = \frac{s}{t} = \frac{2\pi r_e}{T_e}$ $= \frac{2\pi(4.38 \times 10^9)}{6.10 \times 24 \times 3600} = 5.22 \times 10^4 \text{ m s}^{-1} = 52.2 \text{ km s}^{-1}$	2	1			3
c)	<p>According to Kepler's 3rd Law:</p> $\frac{T_1^2}{r_1^3} = \frac{T_2^2}{r_2^3} \quad \left(T = \frac{s}{v} = \frac{2\pi r}{v} \Rightarrow T^2 = \frac{4\pi^2 r^2}{v^2} \right)$ $\frac{4\pi^2 r_1^2}{r_1^3 v_1^2} = \frac{4\pi^2 r_2^2}{r_2^3 v_2^2} \Rightarrow r_1 v_1^2 = r_2 v_2^2 \Rightarrow \left(\frac{v_1}{v_2} \right)^2 = \frac{r_2}{r_1}$ $\Rightarrow \frac{v_1}{v_2} = \sqrt{\frac{r_2}{r_1}}$ <p>Alternative solution: Gravitational force = Centripetal force</p> $ F_G = F_C \Leftrightarrow G \frac{M m}{r^2} = \frac{m v^2}{r} \Rightarrow v^2 = \frac{G M}{r}$ $\Rightarrow \frac{v_1^2}{v_2^2} = \frac{\frac{G M}{r_1}}{\frac{G M}{r_2}} = \frac{r_2}{r_1} \Rightarrow \frac{v_1}{v_2} = \sqrt{\frac{r_2}{r_1}}$		1	2		3

4. Marking Scheme and 5. Answer Sheet

Solution to Question 1, Part A		Fields				
d)	Using the equation from part c), and comparing with planet “e”, whose speed is known from part b), gives:			0.5	3	
	$\frac{v_e}{v_x} = \sqrt{\frac{r_x}{r_e}} \Rightarrow r_x = r_e \left(\frac{v_e}{v_x} \right)^2$ $\Rightarrow r_x = 4.38 \times 10^9 \left(\frac{5.22 \times 10^4}{4.13 \times 10^4} \right)^2$ $\Rightarrow r_x = 7.00 \times 10^9 \text{ m} = 7.00 \times 10^6 \text{ km}$ <p>It is planet “g”.</p>	 <pre> 1.1 1A d RAD ve:=5.22·10^4·_m·_s^-1 ▶ 52200·_m _s re:=4.38·10^6·_km ▶ 4.38E9·_m vx:=4.13·10^4·_m·_s^-1 ▶ 41300·_m _s r:=right(solve(vx=√(re/rx),rx)) ▶ 6.9971E9·_m r▶_km ▶ 6.9971E6·_km </pre>	1	1		0.5
e) i.	$E_{\text{tot}} = E_{\text{kin}} + E_{\text{pot}} = \frac{1}{2}mv^2 - G\frac{mM}{r}$ <p>Since $F_G = F_C \Rightarrow G\frac{mM}{r^2} = \frac{mv^2}{r} \Rightarrow v^2 = \frac{GM}{r}$</p> $\Rightarrow E_{\text{tot}} = E_{\text{kin}} + E_{\text{pot}} = \frac{GMm}{2r} - G\frac{mM}{r} = -G\frac{mM}{2r}$		1		3	
ii.	$E_{\text{tot}} = -G\frac{mM}{2r}$ $E_{\text{tot}} = -6.67 \times 10^{-11} \times \frac{0.77 \times 5.97 \times 10^{24} \times 1.77 \times 10^{29}}{2 \times 4.38 \times 10^9}$ $E_{\text{tot}} = -6.20 \times 10^{33} \text{ J}$		1		1	
		4	7	3	2	16

4. Marking Scheme and 5. Answer Sheet

		Solution to Question 1, Part B		Fields							
		A: Knowledge and Comprehension; B: Application; C: Analysis and Evaluation; W: Written Communication					A	B	C	W	Σ
a)	i.	<p>The centripetal force required is supplied by the magnetic field.</p> $F_m = F_C$ $\Rightarrow Be v = \frac{m_p v^2}{R}$ $\Rightarrow R = \frac{m_p v}{e B}$				1.5				3	
	ii.	<p>The distance travelled during the time Δt is half the circumference of a circle having radius R:</p> $v = \frac{\Delta s}{\Delta t} = \frac{\pi R}{\Delta t} \Rightarrow \Delta t = \frac{\pi R}{v}$ <p>Inserting R from i. gives $\Delta t = \frac{\pi m_p}{e B}$. Hence Δt is independent of the speed v.</p>		2					2		
b)	i.	<p>The potential energy of a charge e in an electric field of potential difference U is $E_{\text{pot}} = e U$.</p> <p>During each gap crossing the potential energy is converted into kinetic energy. Hence, the increase in kinetic energy of the proton for each gap crossing is $\Delta E_{\text{kin}} = e \times 1.00 \times 10^4 \text{ V} = 1.00 \times 10^4 \text{ eV}$.</p>		1.5			0.5		2		
	ii.	<p>For the first gap crossing the initial value of the kinetic energy of the proton is 0. $\Rightarrow \Delta E_{\text{kin}} = E_{\text{kin}} - 0 = E_{\text{kin}}$</p> $E_{\text{kin}} = E_{\text{pot}} \Rightarrow \frac{1}{2} m_p v^2 = e U$ $\Rightarrow v = \sqrt{\frac{2 e U}{m_p}}$ <p>Inserting this formula for v into the equation from a) i. gives</p> $R_1 = \frac{m_p v_1}{e B} = \sqrt{\frac{m_p^2}{e^2 B^2} \times \frac{2 e U}{m_p}}$ $= \sqrt{\frac{2 m_p U}{e B^2}}$ $= \sqrt{\frac{2 \times 1.67 \times 10^{-27} \times 1 \times 10^4}{1.60 \times 10^{-19} \times 1}}$ $= 0.0144 \text{ m} = 1.44 \text{ cm}$			1			2	3		

4. Marking Scheme and 5. Answer Sheet

Solution to Question 1, Part B			Fields									
c)	i.	<p>Using the equation in a) i:</p> $v_{\max} = \frac{eBR_{\max}}{m_p} \text{ and}$ $E_{\text{kin max}} = \frac{1}{2}mv_{\max}^2 \text{ we get}$ $E_{\text{kin max}} = \frac{1}{2}m_p \left(\frac{eBR_{\max}}{m_p} \right)^2$ $E_{\text{kin max}} = \frac{(1.6 \times 10^{-19} \times 1 \times 0.289)^2}{2 \times 1.67 \times 10^{-27}}$ $= 6.40 \times 10^{-13} \text{ J}$ $= 4.00 \times 10^6 \text{ eV} = 4.00 \text{ MeV}$						2				3
	ii.	<p>Dividing the maximum kinetic energy from c) i. by the energy received during each gap crossing (see b) i.), we get the number of gap crossings n.</p> $n = \frac{4.00 \times 10^6 \text{ eV}}{1.00 \times 10^4 \text{ eV}}$ $= 400 \text{ gap crossings,}$ <p>i.e. 200 revolutions</p>						1				1
			6.5	4.5	2.5	0.5	14					

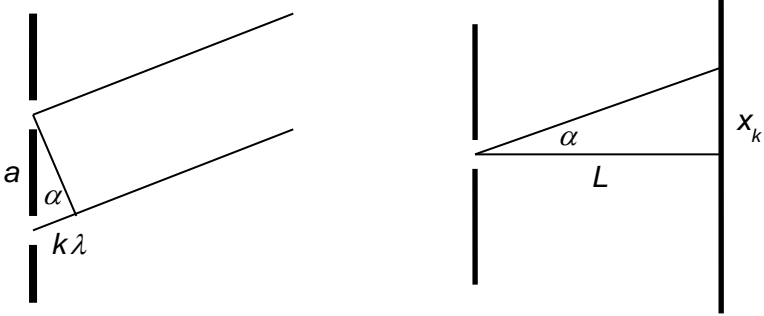
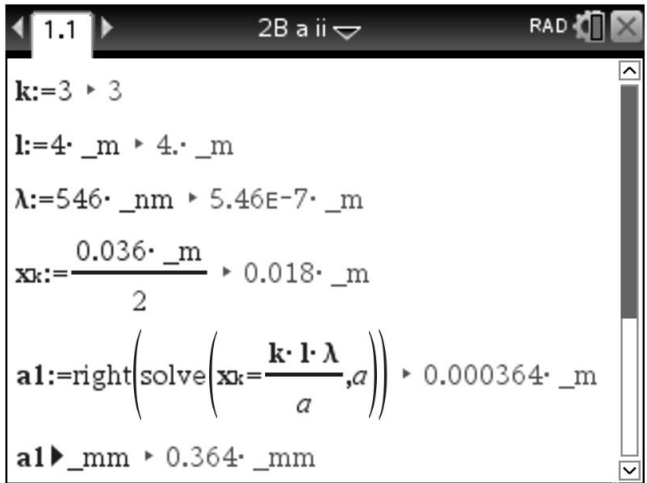
4. Marking Scheme and 5. Answer Sheet

Solution to Question 2, Part A				A	B	C	W	Σ
A: Knowledge and Comprehension; B: Application; C: Analysis and Evaluation; W: Written Communication								
a)	i.		<p style="text-align: center;">open at both ends</p>	<p style="text-align: center;">closed at one end</p>	1			4
		fundamental			1			
		1 st Overtone			1			
(The arrows show the positions of the nodes.)								
	ii.	<p>The wavelength of a sound with frequency 20 Hz is</p> $c = \lambda f \Rightarrow \lambda = \frac{c}{f} = \frac{346}{20} = 17.3\text{m}$ <p>The length of an open pipe is half the fundamental wavelength: $L_{\text{open}} = 8.65\text{ m.}$</p> <p>The length of a closed pipe is a one quarter of the fundamental wavelength: $L_{\text{closed}} = 4.325\text{ m.}$</p>			2			3
	iii.	<p>From the sketches in a) i., for the same L, one can see the relation between the wavelengths for the first overtones for both types of pipes</p> $\frac{\lambda_{\text{open},1}}{\lambda_{\text{closed},1}} = \frac{L}{\frac{4}{3}L} = \frac{3}{4} \text{ and since } f \propto \frac{1}{\lambda} \Rightarrow \frac{f_{\text{open},1}}{f_{\text{closed},1}} = \frac{4}{3}$				2		2
<p>Alternative solution:</p> <p>You could also get this ratio using the formulae:</p> $\lambda_{\text{open},n} = \frac{2L}{n+1} \text{ and } \lambda_{\text{closed},n} = \frac{4L}{2n+1}, \text{ where } n = 0,1,2,3,\dots$ <p>When $n = 1$: $\frac{\lambda_{\text{open},1}}{\lambda_{\text{closed},1}} = \frac{2L}{2} \div \frac{4L}{3} = \frac{3}{4} \Rightarrow \frac{f_{\text{open},1}}{f_{\text{closed},1}} = \frac{4}{3} \text{ since } f \propto \frac{1}{\lambda}$</p>								

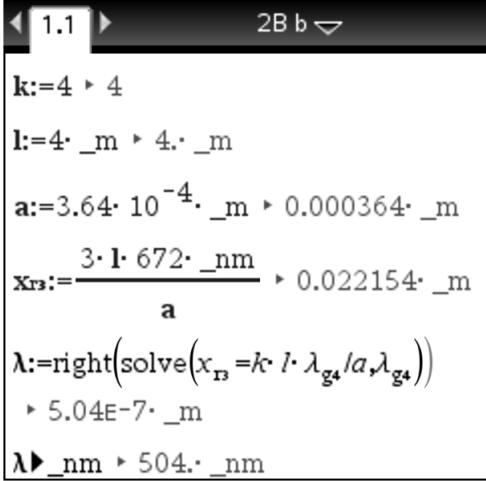
4. Marking Scheme and 5. Answer Sheet

Solution to Question 2, Part A							
b)	i.	The frequency of the note has to be divided by 2 four times. $\frac{440\text{Hz}}{2^4} = 27.5\text{Hz} > 20\text{Hz}$. Hence the human ear can hear this note.	2				2
	ii.	1.	$\frac{14\ 080}{440} = 32 = 2^5$, where 5 is the number of octaves above 440 Hz.	1			1
		2.	The wavelength of the sound with $f = 14\ 080$ Hz is: $\lambda = \frac{c}{f} = \frac{346\text{m s}^{-1}}{14\ 080\text{s}^{-1}} = 0.02457\text{m} = 24.57\text{mm}.$ $\frac{\lambda}{L} = \frac{24.57}{6.14} = 4.$ Since $\lambda = 4L$, it is a closed pipe.		2	1	
			8	4	3	0	15

4. Marking Scheme and 5. Answer Sheet

Solution to Question 2, Part B		A	B	C	W	Σ
A: Knowledge and Comprehension; B: Application; C: Analysis and Evaluation; W: Written Communication						
a) i.	 <p>For constructive interference to occur at a point on a screen, the path difference between the waves meeting at that point must be equal to a whole number multiple of the wavelength of the light.</p> $\frac{k \lambda}{a} = \sin \alpha \approx \tan \alpha = \frac{x_k}{L} \Rightarrow \frac{k \lambda}{a} = \frac{x_k}{L} \Rightarrow x_k = k \frac{L \lambda}{a}$ <p>The following approximations are used:</p> <p>Since $L \gg a$, we can approximate rays from adjacent slits to the screen as being parallel to each other and at an angle α to the central axis.</p> <p>Since α is very small ($a \gg \lambda$), $\sin \alpha \approx \tan \alpha$</p>	1				4
			0.5			0.5
		0.5			0.5	
		0.5			0.5	
ii.	$x_k = k \frac{L \lambda}{a}$ $\Rightarrow a = \frac{k L \lambda}{x_k}$ $= \frac{3 \times 4 \times 546 \times 10^{-9}}{0.036/2}$ $= 3.64 \times 10^{-4} \text{ m}$ $= 0.364 \text{ mm}$ 		1			2
			1			

4. Marking Scheme and 5. Answer Sheet

		Solution to Question 2, Part B				
b)	$x_k = k \frac{L \lambda}{a} \Rightarrow x_{3r} = \frac{3 \times 4.00 \times 672 \times 10^{-9}}{3.64 \times 10^{-4}} = 22.15 \text{ mm}$ <p>Because the green maximum must also be at x_{3r}, $\Rightarrow x_{k\text{green}} = 22.15 \text{ mm}$</p> <p>As $\lambda_{\text{green}} < \lambda_{\text{red}}$ it is sufficient to start with the 4th order green maximum:</p> $x_k = k \frac{L \lambda}{a} \Rightarrow \lambda = \frac{x_k a}{kL} \Rightarrow \lambda = \frac{22.15 \times 10^{-3} \times 3.64 \times 10^{-4}}{4 \times 4.00} = 504 \times 10^{-7} \text{ mm}$ <p>This wavelength $504 \times 10^{-7} \text{ mm}$ is in the range given for green light. Hence the 3rd red maximum overlaps the 4th green maximum. Using $k \geq 5$ gives wavelengths lower than the range given for green light.</p>	0.5			4	
			0.5	1.5	1	
c) i.	$d = \frac{1}{\text{number of lines per metre}} = \text{the grating constant}$ <p>θ_k is the angle between the light rays forming the central maximum and the light rays forming the kth order maximum.</p>			0.5	1	
				0.5		
ii.	$\tan \theta_1 = \frac{x_1}{L} \Rightarrow \theta_1 = \tan^{-1} \frac{x_1}{L} = \tan^{-1} \frac{0.871}{4.00}$ $d = \frac{1}{4000 \times 10^2} \text{ m}^{-1}$ $\lambda = \frac{d \sin \theta}{k} = \frac{1}{4000 \times 10^2} \times \sin \left(\tan^{-1} \frac{0.871}{4.00} \right) = 5.32 \times 10^{-7} \text{ m}$	2			4	
			1			
				1		
		4	6	2	3	15

4. Marking Scheme and 5. Answer Sheet

		Solution to Question 3						
A: Knowledge and Comprehension; B: Application; C: Analysis and Evaluation; W: Written Communication		A	B	C	W	Σ		
a)	i.	<p>hf: the energy of a photon incident on the photoelectric cell</p> <p>W_0: the work function, i.e. the minimum energy needed to release an electron from the surface of the photocathode</p> <p>E_{kin}: the maximum kinetic energy of the released electron</p>	1.5			1.5	3	
	ii. 1.	$E_{\text{photon}} = hf = \frac{hc}{\lambda}$ $= \frac{6.63 \times 10^{-34} \times 3.00 \times 10^8}{486 \times 10^{-9}}$ $= 4.09 \times 10^{-19} \text{ J}$	<pre> 1.1 3 a ii RAD wo:=2.08*_eV 3.3325E-19*_J λ:=486*_nm 4.86E-7*_m a:=100*_mm^2 0.0001*_m^2 f:=right(solve(_c=λ*_f,f)) 6.1686E14*_Hz e_ph:=_h*_f 4.0873E-19*_J ... </pre>		3			3
	2.	<p>Part of the energy of the photon is used to set the electron free. The remainder of the photon's energy is the kinetic energy of the free electron.</p> $E_{\text{kinmax}} = E_{\text{photon}} - W = hf - W$ $E_{\text{kinmax}} = 4.09 \times 10^{-19} - 2.08(1.60 \times 10^{-19})$ $= 7.62 \times 10^{-20} \text{ J } (= 0.476 \text{ eV})$	<pre> 1.1 3 a ii ... ekin:=eph-wo 7.5481E-20*_J ... </pre>		1	1		2
	3.	<p>n is the number of photons incident on the photocathode per second.</p> $P = n E_{\text{photon}} \Rightarrow n = \frac{P}{E_{\text{photon}}}$ $= \frac{1 \times 10^{-1} \times 100 \times 10^{-6}}{4.09 \times 10^{-19}}$ $= 2.44 \times 10^{13} \text{ s}^{-1}$	<p>...</p> $p:=0.1 \cdot \frac{W}{m^2} \cdot a \rightarrow 0.00001 \cdot W$ $n:=\frac{P}{e_{ph}} \rightarrow 2.4466E13 \cdot \text{Hz}$ <p>...</p>		1		2	4
4.	<p>Only 4% of the incident photons release an electron.</p> <p>n' is the number of photons which actually cause photoemission per second.</p> $n' = \frac{4}{100} \times 2.44 \times 10^{13} \text{ s}^{-1} = 9.76 \times 10^{11} \text{ s}^{-1}$ $I = \frac{Q}{t} = e n' = \frac{1.60 \times 10^{-19} \times 9.76 \times 10^{11}}{1}$ $= 1.56 \times 10^{-7} \text{ A}$	<p>...</p> $i:=0.04 \cdot n \cdot q \rightarrow 1.5679E-7 \cdot \text{A}$		2		2	4	

4. Marking Scheme and 5. Answer Sheet

Solution to Question 3						
b)	<p>In a) ii. 2. the energy of a photon with wavelength 486 nm is given as</p> $E_{\text{photon}} = 4.09 \times 10^{-19} \text{ J} = \frac{4.09 \times 10^{-19}}{1.60 \times 10^{-19}} \text{ eV} \approx 2.56 \text{ eV}.$ $\Delta E = -3.40 - E_n = -2.56$ $\Rightarrow E_n = -3.40 + 2.56 = -0.84 \text{ eV} \Rightarrow n = 4$ <p>Hence the transition is from level 4 to level 2.</p>	2	1.5	0.5	4	
		6.5	7.5	4	2	20

4. Marking Scheme and 5. Answer Sheet

		Solution to Question 4				
A: Knowledge and Comprehension; B: Application; C: Analysis and Evaluation; W: Written Communication		A	B	C	W	Σ
a)	i.	The atoms of each isotope of an element have the same number of protons (i.e. the same atomic number) but have a different number of neutrons (i.e. a different mass number).				1
	ii.	${}_{43}^{99}\text{Tc}$ has 43 protons and $99 - 43 = 56$ neutrons.				0.5
	iii.	${}_{43}^{99}\text{Tc} \rightarrow {}_{44}^{99}\text{Ru} + {}_{-1}^0\text{e} + \left({}_{0}^{0-}\nu\right) = {}_{44}^{99}\text{Ru} + \beta^{-} + \left({}_{0}^{0-}\nu\right)$. It is β^{-} decay. (The anti-neutrino need not be given by students.)				1
	iv.	The half-life of a radioactive isotope is the time taken for the number of radioactive nuclei N to fall to half of the original value N_0 .				0.5
	v.	The graph starts at time $t = 0$ with an activity of 4000 Bq and falls to 1000 Bq after about 12 h. So the half-life is 6 h (approximately). (Other points of the graph may be chosen.)				0.5
	vi.	$\frac{1}{2}N_0 = N_0 e^{-\lambda T_{1/2}} \Rightarrow \frac{1}{2} = e^{-\lambda T_{1/2}} \Rightarrow \ln \frac{1}{2} = -\lambda T_{1/2} \Rightarrow -\ln 2 = -\lambda T_{1/2} \Rightarrow T_{1/2} = \frac{\ln 2}{\lambda}$				2
b)	i.	A chain reaction occurs when at least one neutron from each fission reaction causes further fission to produce more fission neutrons, so that the fission reaction is self-sustaining.				1
		The neutrons released by the fission of ${}^{235}\text{U}$ are fast neutrons and must be slowed down in order to cause further fission in ${}^{235}\text{U}$ which is only fissionable by slow neutrons. Moderators e.g. heavy water or graphite, slow down the fast neutrons to speeds which allow them to cause fission of ${}^{235}\text{U}$ atoms.				1.5
	ii.	Mass of the reactants - Mass of the products = Δm $[m({}_{92}^{235}\text{U}) - 92m_e + m_n] - [m({}_{56}^{141}\text{Ba}) - 56m_e + m({}_{36}^{92}\text{Kr}) - 36m_e + 3m_n] = \Delta m$ (If the student explains that the total number of electrons does not change, it is not necessary to mention them in the above equation.) $\Delta m = m({}_{92}^{235}\text{U}) - m({}_{56}^{141}\text{Ba}) - m({}_{36}^{92}\text{Kr}) - 2m_n$ $\Delta m = 235.043930 - 140.914411 - 91.926156 - 2(1.008665)$ $\Delta m = 0.186033 \text{ u}$ $\Delta m = 0.186033 \times 931.5 \frac{\text{MeV}}{\text{c}^2} = 173.3 \frac{\text{MeV}}{\text{c}^2}$ $\Delta E = \Delta mc^2 = 173.3 \text{ MeV} = E_{\text{fission}}$				2

4. Marking Scheme and 5. Answer Sheet

Solution to Question 4								
c)	$2 \text{ GW} = 2 \frac{\text{GJ}}{\text{s}}$ $= 2 \times 3600 \text{ GJ per hour}$ $= 7.20 \times 10^{12} \text{ J per hour}$ <p>The number of fission reactions needed per hour, n, (taking into account the efficiency), is:</p> $n = \frac{7.2 \times 10^{12}}{E_{\text{fission}} \times 0.33}$ $= \frac{7.2 \times 10^{12}}{210 \times 10^6 \times 1.6 \times 10^{-19} \times 0.33}$ $= 6.49 \times 10^{23} \text{ fissions per hour}$ <p>The mass needed per fission reaction is:</p> $m_1 = 235 \text{ u} = 235 \times 1.66 \times 10^{-27} \text{ kg}$ $= 3.90 \times 10^{-25} \text{ kg}$ <p>The total mass needed per 1 hour:</p> $m = m_1 \times n = 3.90 \times 10^{-25} \times 6.49 \times 10^{23}$ $= 0.253 \text{ kg per hour.}$	<p>1.1 4 c ▾</p> <p>ex:=210· 10⁶· _eV ▶ 3.3646E-11· _J</p> <p>etot:=2· 10⁹· _W ▶ 2.E9· _W</p> <p>n:=$\frac{etot}{ex \cdot 0.33}$ ▶ 1.8013E20· _Hz</p> <p>m:=n· 235· _amu ▶ 0.00007· $\frac{\text{kg}}{\text{s}}$</p> <p>m▶_kg· _hr⁻¹ ▶ 0.25305· $\frac{\text{kg}}{\text{hr}}$</p>	1	1	1			
				7	7	3.5	2.5	20

